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The isolation from the epigeal part of <u>Vicia balansae</u> of five compounds has been reported previously. The structures of four of them were described [1].

We now give the results of an investigation of the fifth compound. The air-dry raw material (1 kg) was exhaustively extracted with 60% ethanol. The extract, concentrated to 1/3 of its original volume, was treated successively with chloroform, ethyl acetate, and n-butanol. On concentration, the butanolic fraction deposited a precipitate containing six flavonoid compounds of which the main one proved to be substance (V). For its isolation, the precipitate was deposited on a column of polyamide sorbent and elution was performed successively with water and increasing concentrations of aqueous ethanol.

Substrate (V) for med pale yellow crystals with mp 204-209°C,  $[\alpha]_D^{20}$  =63.88 ±2° (c 1.34; pyridine); Rf 0.15 in the butan-1-ol-CH<sub>3</sub>COOH-H<sub>2</sub>O (4:1:5) system and 0.10 in 15% CH<sub>3</sub>COOH.  $\lambda_{max}$ , nm: CH<sub>3</sub>OH-269, 238; CH<sub>3</sub>COONa-268, 345; 387, 390; H<sub>3</sub>BO<sub>3</sub> + CH<sub>3</sub>COONa = 269, 342; AlCl<sub>3</sub> = 278, 348, 383; AlCl<sub>3</sub> + HCl = 255, 277, 346, 382; CH<sub>3</sub>ONa = 259, 394 nm. According to the PMR spectroscopy of the acetyl derivative in CDCl<sub>3</sub>, the substance was a diglycoside attached to the aglycone by a  $\beta$ -glycosidic bond. In the 1.98-2.24 ppm region ( $\delta$  scale) there were the signals of 10 acetyl groups (3 OH), and at 4.14 and 5.20 ppm the signals of twelve H atoms of glucose residues. The protons of aromatic rings gave signals at 6.41, 7.15, and 7.51 ppm (H-6, 8; H-5; H-2, 6). Quantitative acid hydrolysis confirmed that the compound was a diglycoside. There were two moles of D-glucose per one mole of aglycone.

The aglycone formed pale yellow crystals with mp 330°C, M<sup>+</sup> 286. It was identified from its physico-chemical characteristics (UV and mass spectroscopy, R<sub>f</sub>, and the products of alkaline hydrolysis) as luteolin.

The results of the study of the products of stepwise hydrolysis (0.1 N HCl, 120 min, 100°C) showed that substance (V) was luteolin 4',7-di-O- $\beta$ -D-glucopyranoside.

## LITERATURE CITED

1. O. A. Andreeva, Khim. Prir. Soedin., 720 (1980).

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